Code No.	J – 2267
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Entra	anc	e Examination	on for A		sion to tments		G. Cou	rses in	the Te	aching
					CSS					
				BIOTE	CHNO	LOGY				
				Gener	al Instru	ctions				
		Question Pape riptive type (40		ing two	Parts —	Part 'A	' Objecti	ve type	(60%) 8	Part 'B'
	•	ctive type ques			•			e (✔) 'tio	ck marke	ed' in the
3. 8	3 que	estions are to b	e answe	red out o	of 12 que	estions o	carrying 5	5 marks	each in F	Part 'B'.
		ative marking ort 'A'.	: 0.2	5 marks	s will b	e dedi	ucted fo	or each	wrong	answer
Time :	: 2 H	lours						N	Max. Mar	ks : 100
To be	e fille	ed in by the Car	ndidate							
Regis		in Figures								
Numb	ber	in words								
				-	ADT	•				
					PART – A jective Ty					
01				` -	•	. ,				
Choose	е ар	propriate answ	er from t	he optio	ns in the	questio	ns. One) marks)
1. T	he p	orimary hosts t	for HIV-	1 virus a	are					
a) -	T8 killer cells			b)	B cel	ls			
C)) -	T4 helper cells	8		d)	Mond	ocytes			

DONOTWRITEHERE

- 2. Which of the following amino acids contain sulphur?
 - a) Tyrosine

b) Methionine

c) Alanine

- d) Glutamic acid
- 3. Placental connection is typical of
 - a) Viviparous reproduction
- b) Ovoviparous reproduction
- c) Oviparous reproduction
- d) All of the above
- 4. Prokaryotic ribosomes are
 - a) 40S

b) 70S

c) 80S

d) 58S

5.	Mal	tose is composed of		
	a)	Galactose + Glucose	b)	Glucose + Fructose
	c)	Glucose + Glucose	d)	Galactose + Ribose
6.	Rad	lula is a structure used for food inta	ake by	/
	a)	Arthropods	b)	Nematodes
	c)	Annelids	d)	Molluscs
7.	Whi	ch kind of cleavage is shown by ma	amma	alian embryos?
	a)	Rotational	b)	Radial
	c)	Spiral	d)	Bilateral
8.	A th	ree chambered heart is found in		
	a)	Mammals except human beings	b)	Birds
	c)	Amphibians	d)	Fishes
9.	Ribo	oflavin is another name for		
	a)	Vitamin B2	b)	Vitamin B5
	c)	Vitamin A	d)	Vitamin B12
10.	If pl	H of a solution is 5.6, then its pOH	is	
	a)	7.4	b)	8.2
	c)	8.4	d)	7.6
11.		tein 'X' from <i>E.coli</i> contains 125 sent in the gene encoding the prote		o acids. The number of nucleotides
	a)	375	b)	388
	c)	378	d)	381
12.	For	an exergonic reaction		
	a)	$\Delta H > 0$	b)	$\Delta G > 0$
	c)	$\Delta S > 0$	d)	All of the above

13.	Stap a) b) c) d)	The thick peptidoglycan layer retains the crystal violet stain washes of the outer lipopolysaccharide layed. The crystal violet stains and the sains are the	ins the and are reta	ne crystal violet stain safranin counterstain is seen
14.	Bac	teria use which of the following org	anell	es for locomotion
	a)	Pili	b)	Flagella
	c)	Ribosome	d)	Capsule
15.		ins are commonly used as inhibi	tors	of HMG CoA reductase, this kind of
	a)	Uncompetitive inhibition	b)	Non competitive inhibition
	c)	Competitive inhibition	d)	Mixed inhibition
16.	nor	his parents have the same proble	m. H	I and green, however neither his sister is mother's father also wasn't able to he inheritance pattern of this disorder?
	a)	Autosomal dominant colour blindr	ness	
	b)	X-linked recessive colour blindnes	SS	
	c)	Autosomal recessive colour blindr	ness	
	d)	X-linked dominant colour blindnes	ss	
17.		growth of a bacteria can be plotte	ed or	n a graph in the form of a curve. The
	a)	Stationary phase	b)	Lag phase
	c)	Log phase	d)	End of lag phase
18.		ole sulfur bacteria is a group of phore		nthetic bacteria that are different from
	a)	They are purple in colour		
	b)	They require red light to carry out	phot	osynthesis
	c)	They use sulfur as electron donor	for p	hotosynthesis
	d)	They produce oxygen via photosy	nthe	sis

19.	Prol	karyotes DNA polymerase involved	in re	moving RNA primer
	a)	DNA Pol I	b)	DNA Pol II
	c)	DNA Pol III	d)	DNA Pol α
20.		ch one of the following is a nitrocciation with leguminous plants?	ogen	fixing bacteria seen in a symbiotic
	a)	Frankia	b)	Acetobacter
	c)	Azospirillium	d)	Rhizobium
21.	Botu	ulism is a microbial infection acquire	ed by	,
	a)	Mosquito bite	b)	Needle prick injury
	c)	Contamination of food	d)	Aerosol containing spores
22.	Duri	ing the annealing step in a PCR re	actio	n
	a)	The double stranded dna is separ	ated	
	b)	The primers bind to the template s	strand	d
	c)	dNTPs are added to the primer		
	d)	the polymerase enzyme becomes	activ	e and catalyses the reaction
23.	Trar	nsfer of genetic material between tw	vo ba	acteria via sex pili is called
	a)	conjugation	b)	transformation
	c)	transduction	d)	gene recombination
24.	A m	ammogram is a routine screening t	est fo	or breast carcinoma. It is basically a
	a)	Biopsy of breast tissue	b)	Ultrasound examination of breast
	c)	X-ray of the breast	d)	CT scan of the breast
25.	Whi	ch of the following is less condense	ed po	rtion of chromatin?
	a)	Pseudochromatin	b)	Multichromatin
	c)	Heterochromatin	d)	Euchromatin

26.	Mar	k the process which does not invol	ve hi	stone modification.
	a)	Acetylation	b)	Methylation
	c)	Phosphorylation	d)	Dehydration
27.	The	type of plasmid which make the h	ost ce	ell resistant to antibiotic
	a)	R plasmid	b)	F plasmid
	c)	Col plasmid	d)	Hfr
28.		me the type of bacteria which us bon for growth	es c	arbon dioxide as the sole source of
	a)	Autotroph	b)	Chemotroph
	c)	Organotroph	d)	Phototroph
29.		e of culture which is prepared by anism to culture media	inoc	culating directly from the tissue of an
	a)	Cell line	b)	Transformed cells
	c)	Primary culture	d)	Secondary culture
30.	Enz	zyme which unwinds the double hel	ix of	DNA is
	a)	Topoisomerase	b)	Ligase
	c)	Helicase	d)	Polymerase
31.	Whi	ich of the following is also known a	s jum	ping genes?
	a)	Cistron	b)	Transposon
	c)	Exon	d)	Replicon
32.	The	only woman to win an unshared n	obel _l	prize in physiology and medicine
	a)	Barbara McClintoch	b)	Marie Curie
	c)	Rosalind Franklin	d)	Elizabeth H Blackburn
33.	Whi	ich of the following is not a type of	cance	er?
	a)	Myeoloma	b)	Sarcoma
	c)	Leukemia	d)	Condyloma

34.	Cho	oose a repressible operon from the	follo	wing?
	a)	Lac operon	b)	Gal operon
	c)	Trp operon	d)	None of the above
35.	Wh	ich one of the following infections is	s not	transmitted by mosquitoes?
	a)	Filaria	b)	Kala-azar
	c)	Japanese Encephalitis	d)	Influenza
36.	Car	popedal spasm is a characteristic s	symp	tom seen in which of the following?
	a)	Scarlet fever	b)	Typhoid
	c)	Tetanus	d)	Black water fever
37.	Wh	ich of the following is a routine scre	enin	g test for cervical carcinoma?
	a)	Amniocentesis	b)	Chorionic villi sampling
	c)	Pap Smear	d)	Colonoscopy
38.	Xer	ophthalmia is a known vitamin defi	cienc	y disorder which may be alleviated by
	a)	Sunflower seeds	b)	Mushrooms
	c)	Carrots	d)	Lentils
39.	Wh	ich of the following is a defensive n	necha	anism of innate immunity system?
	a)	Hair	b)	Nails
	c)	Epithelium	d)	Sweat
40.	Ery	throblastosis Foetalis occurs if		
	a)	Mother is Rh + and the fetus is R	h-	
	b)	Mother is Rh- and fetus is Rh+		
	c)	Mother is Rh- and Fetus is Rh- second pregnancy	+ in f	irst pregnancy, while fetus is Rh+ in
	d)	None of the above		

41.	Wha	at is the absorption maxima of Pho	tosys	tem I?
	a)	700 nm	b)	680 nm
	c)	670 nm	d)	710 nm
42.		bodies which can cross placenta pectively are	a and	d are involved in allergic reactions,
	a)	IgG and IgA	b)	IgM and IgE
	c)	IgG and IgE	d)	IgD and IgM
43.		interaction primarily important fo	or the	e stabilization of tertiary structure of
	a)	Hydrogen bond	b)	Hydrophobic interactions
	c)	Wander-waals interactions	d)	Disulphide linkages
44.		ch among the following amino siological pH?	acid	has greatest buffering capacity at
	a)	Tyrosine	b)	Alanine
	c)	Histidine	d)	Cysteine
45.		oidal solution in which disperse ersion medium is called?	ed ph	nase has very little affinity for the
	a)	Hydrophobic colloids	b)	Lysogenic colloids
	c)	Lysophobic colloids	d)	None of the above
46.	Wha	at is isoelectric pH?		
	a)	pH at which a molecule has no ch	arge	
	b)	pH at which a protein is stable		
	c)	pH at which net charge of a mole	cule is	s zero
	d)	pH at which the molecule can be	sepai	rated
47.		•		a semi-permeable membrane that two sides of the membrane is called?
	a)	Potential difference	b)	Donnan equilibrium
	c)	Osmotic potential	d)	Equilibrium

48.	Whi	ch among the following is an exam	ple o	f passive transport?
	a)	Transport of glucose across mem	brane	e using ATP
	b)	Osmosis		
	c)	Sodium Potassium pump		
	d)	Uptake of glucose in root hair		
49.	Opt	ically inactive amino acid is?		
	a)	Tyrosine	b)	Proline
	c)	Glycine	d)	Alanine
50.	Wha	at is the principle of spectrophotom	eter?	
	a)	Refraction of light	b)	Beer-Lambert's law
	c)	Defraction	d)	Huygens-Fresnel principle
51.	Cho	ose the correct statement about S	DS?	
	a)	SDS is an anionic detergent that	stabili	ize the protein
	b)	SDS is a cationic detergent that g	ives	uniform charge to protein
	c)	SDS breaks the disulphide bridge	S	
	d)	SDS is an anionic detergent that	gives	uniform charge to protein
52.	Whi	ch among the following is not a rec	lucino	g sugar?
	a)	Glucose	b)	Sucrose
	c)	Cellobiose	d)	Fructose
53.	Cho	ose the executioner caspase from	the fo	ollowing?
	a)	Caspase 5	b)	Caspase 3
	c)	Caspase 9	d)	Caspase 2
54.	Fat	soluble vitamins are		
	a)	Vitamin A, B12, D	b)	Vitamin D, C, K, E
	c)	Vitamin K, E, D, A	d)	Vitamin B6, D, A, E

55.	Res	spiratory acidosis occurs when		
	a)	the kidneys are not removing acid	d fron	n the body
	b)	the lungs can't remove the carbon	n dio	kide (CO ₂) produced by the body
	c)	increased respiration elevates the	e bloc	od pH beyond the normal range
	d)	increased potassium level		
56.	In th	he presence of 2, 3-Bisphosphogly	cera	te, hemoglobin's affinity to oxygen
	a)	Increases	b)	Decreases
	c)	No effect	d)	Increase at higher temperature
57.	Wh	ere does Beta oxidation takes plac	e in e	eukaryotes?
	a)	Peroxisome	b)	Mitochondria
	c)	Cytosol	d)	Both (a) and (b)
58.	Wh	ich TCA cycle enzyme is also found	d in e	electron transport chain?
	a)	Fumerase		
	b)	Succinate dehydrogenase		
	c)	Isocitrate dehydrogenase		
	d)	Citrate synthase		
59.	Cho	pose the wrong statement about pe	ptide	bond formation
	a)	It is a condensation reaction		
	b)	Peptide bonds are formed between carboxyl group of other amino act		amino group of one amino acid and
	c)	There are 3 peptide bonds in a tri	pepti	de
	d)	Peptide bond formation require A	TP	
60.	RN	A required for protein synthesis		
	a)	mRNA	b)	tRNA
	c)	rRNA	d)	All of the above

10

J - 2267

ANSWER SHEET — PART – A

1 A B C D E 21 A B C D E 41 A B C D E 2 A B C D E 22 A B C D E 42 A B C D E 3 A B C D E 24 A B C D E 44 A B C D E 4 A B C D E 25 A B C D E 44 A B C D E 5 A B C D E 25 A B C D E 44 A B C D E 6 A B C D E 26 A B C D																		
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4 A B C D E 24 A B C D E 44 A B C D E 5 A B C D E 25 A B C D E 45 A B C D E 6 A B C D E 26 A B C D E 46 A B C D E 7 A B C D E 26 A B C D E 46 A B C D E 8 A B C D E 29 A B C D E 48 A B C D E 9 A B C D E 30 A B C D E 50 A B C D E 10 A B C	2	Α	В	С	D	Е	22	Α	В	С	D	Е	42	Α	В	С	D	Ε
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7 A B C D E 27 A B C D E 47 A B C D E 8 A B C D E 28 A B C D E 48 A B C D E 9 A B C D E 29 A B C D E 49 A B C D E 10 A B C D E 30 A B C D E 50 A B C D E 11 A B C D E 31 A B C D E 51 A B C D E 12 A B C D E 32 A B C D E 52 A B C D E 13 A B <t< td=""><td>5</td><td>Α</td><td>В</td><td>С</td><td>D</td><td>Е</td><td>25</td><td>Α</td><td>В</td><td>С</td><td>D</td><td>Е</td><td>45</td><td>Α</td><td>В</td><td>С</td><td>D</td><td>Ε</td></t<>	5	Α	В	С	D	Е	25	Α	В	С	D	Е	45	Α	В	С	D	Ε
8 A B C D E 28 A B C D E 48 A B C D E 9 A B C D E 29 A B C D E 49 A B C D E 10 A B C D E 30 A B C D E 50 A B C D E 11 A B C D E 31 A B C D E 51 A B C D E 12 A B C D E 32 A B C D E 52 A B C D E 13 A B C D E 53 A B C D E 14 A B C D E 34 A B <	6	Α	В	С	D	Е	26	Α	В	С	D	Е	46	Α	В	С	D	Е
9 A B C D E 29 A B C D E 50 A B C D E 11 A B C D E 31 A B C D E 51 A B C D E 12 A B C D E 32 A B C D E 52 A B C D E 13 A B C D E 53 A B C D E 14 A B C D E 54 A B C D E 14 A B C D E 34 A B C D E 55 A B C D E 15 A B C D E 15 A B C D E 15 A B C D E 16 A B C D E 36 A B C D E 16 A B C D E 37 A B C D E 56 A B C D E 17 A B C D E 38 A B C D E 58 A B C D E 18 A B C D E 38 A B C D E 59 A B C D E 19 A B C D E 59 A B C D E	7	Α	В	С	D	Е	27	Α	В	С	D	Ε	47	Α	В	С	D	Ε
10 A B C D E 30 A B C D E 50 A B C D E 11 A B C D E 31 A B C D E 51 A B C D E 12 A B C D E 32 A B C D E 52 A B C D E 13 A B C D E 53 A B C D E 14 A B C D E 34 A B C D E 54 A B C D E 15 A B C D E 15 A B C D E 55 A B C D E 15 A B C D E 16 A B C D E 36 A B C D E 16 A B C D E 36 A B C D E 17 A B C D E 37 A B C D E 57 A B C D E 18 A B C D E 38 A B C D E 58 A B C D E 18 A B C D E 39 A B C D E 59 A B C D E 59 A B C D E	8	Α	В	С	D	Е	28	Α	В	С	D	Е	48	Α	В	С	D	Е
11 A B C D E 31 A B C D E 51 A B C D E 12 A B C D E 32 A B C D E 52 A B C D E 13 A B C D E 13 A B C D E 53 A B C D E 14 A B C D E 34 A B C D E 54 A B C D E 15 A B C D E 15 A B C D E 15 A B C D E 16 A B C D E 36 A B C D E 56 A B C D E 17 A B C D E 37 A B C D E 57 A B C D E 18 A B C D E 38 A B C D E 58 A B C D E 19 A B C D E 39 A B C D E 59 A B C D E	9	Α	В	С	D	Е	29	Α	В	С	D	Е	49	Α	В	С	D	Е
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13 A B C D E 33 A B C D E 53 A B C D E 14 A B C D E 34 A B C D E 54 A B C D E 15 A B C D E 35 A B C D E 55 A B C D E 16 A B C D E 36 A B C D E 56 A B C D E 17 A B C D E 37 A B C D E 57 A B C D E 18 A B C D E 38 A B C D E 58 A B C D E 19 A B	11	Α	В	С	D	Е	31	Α	В	С	D	Е	51	Α	В	С	D	Е
14 A B C D E 34 A B C D E 54 A B C D E 15 A B C D E 55 A B C D E 16 A B C D E 56 A B C D E 17 A B C D E 57 A B C D E 18 A B C D E 58 A B C D E 19 A B C D E 59 A B C D E	12	Α	В	С	D	Е	32	Α	В	С	D	Е	52	Α	В	С	D	Е
15 A B C D E 35 A B C D E 16 A B C D E 36 A B C D E 17 A B C D E 37 A B C D E 18 A B C D E 38 A B C D E 19 A B C D E 39 A B C D E	13	Α	В	С	D	Е	33	Α	В	С	D	Е	53	Α	В	С	D	Е
16 A B C D E 36 A B C D E 56 A B C D E 17 A B C D E 57 A B C D E 18 A B C D E 58 A B C D E 19 A B C D E 59 A B C D E	14	Α	В	С	D	Е	34	Α	В	С	D	Е	54	Α	В	С	D	Е
17 A B C D E 37 A B C D E 57 A B C D E 18 A B C D E 58 A B C D E 19 A B C D E 39 A B C D E 59 A B C D E	15	Α	В	С	D	Е	35	Α	В	С	D	Е	55	Α	В	С	D	Е
18 A B C D E 38 A B C D E 58 A B C D E 19 A B C D E	16	Α	В	С	D	Е	36	Α	В	С	D	Е	56	Α	В	С	D	Е
19 A B C D E 39 A B C D E 59 A B C D E	17	Α	В	С	D	Е	37	Α	В	С	D	Е	57	Α	В	С	D	Е
	18	Α	В	С	D	Е	38	Α	В	С	D	Е	58	Α	В	С	D	Е
20 A B C D E 40 A B C D E 60 A B C D E	19	Α	В	С	D	E	39	Α	В	С	D	Е	59	Α	В	С	D	Е
	20	Α	В	С	D	Е	40	Α	В	С	D	Е	60	Α	В	С	D	Е

BIOTECHNOLOGY

PART – B

(Descriptive Type)

Answer **any eight** questions.

 $(8 \times 5 = 40 \text{ Marks})$

- 1. What is hybridoma technology? Explain the selection process.
- 2. What is the principle of Nuclear Magnetic Resonance?
- 3. What are enantiomers and give one example?
- 4. Which are the important buffers in biological system?
- 5. What is the difference between reducing and non-reducing sugars with examples?
- 6. Define Bohr effect.
- 7. State the difference between oxidative phosphorylation and substrate level phosphorylation.
- 8. Explain the chemiosmotic theory of ATP production.
- 9. What are polysomes?
- 10. Differentiate between isotypic, allotypic and idiotypic and antibodies.
- 11. Differentiate between monoclonal and poyclonal antibodies.
- 12. Name any five plant growth hormones and briefly explain their function.

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